

MSM Trace Element (TE) solution

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Abstract

Trace element solution for Mineral Salts Medium (MSM, see separate protocol

"<http://www.nature.com/protocolexchange/protocols/3809>":<http://www.nature.com/protocolexchange/protocols/3809>). To be added when medium is not supplied with mineral. Add 1 ml of 100x Stock TE solution into 99 ml of MSM.

Reagents

Trace Element solution:

Ferric chloride hexahydrate, $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$

Copper sulfate pentahydrate, $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

Boric acid, HBO_3

Manganese sulfate hydrate, $\text{MnSO}_4 \cdot x\text{H}_2\text{O}$

Sodium molybdate dihydrate, $\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$

Cobalt chloride hexahydrate, $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$

Zink sulfate heptahydrate, $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$

Sodium selenate, Na_2SeO_4

General:

Sulfuric acid

Sodium hydroxide

Equipment

Stirring rods,

Magnetic stirrer,

Measuring colves,

pH meter

Procedure

For 100x Stock TE solution:

1100 mg/l $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$

50 mg/l $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

200 mg/l HBO_3

200 mg/l $\text{MnSO}_4 \cdot \text{xH}_2\text{O}$

80 mg/l $\text{Na}_2\text{MoO}_4 \cdot \text{x2H}_2\text{O}$

60 mg/l $\text{CoCl}_2 \cdot \text{x6H}_2\text{O}$

90 mg/l $\text{ZnSO}_4 \cdot \text{x7H}_2\text{O}$

10 mg/l Na_2SeO_4

Add the constituents into a final volume of 1 liter of water (pH pre-adjusted to around 2.5), one at a time, dissolving before adding the next.

Adjust the pH to 1.5 with sulfuric acid (and/or sodium hydroxide)

-sterile filter

Timing

about 30 minutes

Troubleshooting

-TE solution should be clear and almost colorless. Strong coloration indicates too high pH.

-When encountering difficulties during dissolution of a reagent, try adding some more sulfuric acid.